## THE STRUCTURE OF RAUCAFFRICINE

## M. Ataullah Khan and A. M. Ahsan Karachi Laboratories, Pakistan Council of Scientific and Industrial Research, Off University Road, Karachi-39, Pakistan.

(Received in UK 9 November 1970; accepted for publication 18 November 1970)

For raucaffricine, an alkaloid isolated from Rauwolfia caffra Sonder by one of the present authors, M.A.K., and co-workers<sup>1</sup>, they proposed the molecular formula  $C_{26}H_{32}O_8N_2$ ,  $\frac{1}{2}H_2O$ , for its tetra-acetate  $C_{34}H_{40}O_{12}N_2$ , for its tetrabensoate  $C_{54}H_{48}O_{12}N_2$  and for the methiodide  $C_{26}H_{32}O_8N_2$ ,  $CH_3I$ . In accordance with the structure (I) we now propose, the proof

of which is in the sequel, the molecular formula for the base is modified to  $^{\rm C}_{27}{\rm H}_{32}{\rm O}_8{\rm N}_2$ ,  $^{\frac{1}{2}}_2{\rm H}_2{\rm O}$  and thus the tetraacetate becomes  $^{\rm C}_{35}{\rm H}_4{\rm O}_{12}{\rm N}_2$ , the tetrabenzoate  $^{\rm C}_{55}{\rm H}_4{\rm S}{\rm O}_{12}{\rm N}_2$  and the methiodide  $^{\rm C}_{27}{\rm H}_{32}{\rm O}_8{\rm N}_2$ ,  $^{\rm CH}_3{\rm I}$ , which is in agreement with the analytical data of the above compounds.

The tw spectrum of rancaffricine exhibits absorption in the regions known for indolenine alkaloids,  $\lambda$  max 219.8, 258 mm;  $\lambda$  min 236 mm<sup>1</sup>. Its easy conversion to indole bases, under mild alkaline conditions similar to that of perakine<sup>2</sup> also suggests that it is

an indolenine derivative<sup>3</sup>. Thus we recorded the nur spectrum (60 MHz) of raucaffricine in pyridine and observed a close similarity to that of vomilenine<sup>4</sup>.

Attention is specially drawn to the peaks for the ethylidine group (1.56 5, 3H, d, J=6.7 Hz; 5.985, 2H, q, J=7 Hz), the single hydrogens at C-15 (2.805, 1H, m), C-5 (3.335, 1H, q)

and at C-3 (4.316, 1H). The peak centred at 5.006 corresponds to that in the twedlenine spectrum at exactly the same position. A noteworthy difference in the two spectra is for the anomeric hydrogen atom which appears as a broad doublet at 5.425.

Hydrolysis of rancaffricine with #N hydrochloric acid yielded two components, one of which was identified as D-(+)-galactose through paper chromatography and the preparation of the osazone. The aglycone (as hydrochloride and free base) was identical (ir, uv and tlc) with the product of a parallel experiment with vomilenine run under identical conditions. In both cases uv reveals a complete change from indolemine to indole type bases, as is to be expected.

Ensymmetric studies suggested that the linkage at the anomeric C atom is  $\mathcal{L}_{\bullet}$ . Sweet almond emulsin does not affect rancaffricine, while an attempted partial synthesis from its constituents in presence of brower's yeast shows its formation on the (silica gel) (butanol: acetic acid: water: 4:1:1).

Molecular models of rancaffricine also indicate that the preferred glycosidic linkage is  $_{\rm s}$ , that the C-21 oxygen of the vomilenine moiety is axial and the geometry of the C-15 methyl is possibly that of the sarpagine alkaloids  $^{7}(I)$ .

Acknowledgements We are indebted to Messrs, Bayer-Pharma, (Pakistan) Ltd. for the supply of fresh brewer's yeast and to Dr. A. Hofmann of Sandos A.-G., Basle, Sudtserland for a sample of vomilenine.

## References

- 1. N.H. Khen, M. Ataullah Khen, and S. Siddiqui, Pakistan J. Sci. Ind. Res., 8, 23 (1964).
- P.R. Wishafer, M.F. Bartlett, L. Dorfmen, M.A. Gillen, E. Schlittler and Ernest Wenkert, Tetrahedron, 363 (1961).
- 3. N.H. Khan, M. Ataullah Khan, and S. Siddiqui, Pakistan J. Sci. Ind. Res., 9, 210 (1966).
- 4. W.I. Taylor, A.J. Frey, and A. Hofmann, Helv. Chim. Acta, 45, 611 (1962).
- 5. Compare position 5.428 for similar proton of (-)-veronamine. M. Shamma, M.G. Kelly, and Sr. M.A. Podcsasy, Tetrahedron Letters, 4951 (1969).
- 6. B.D.E. Gaillard, Nature, 171, 1160 (1953).
- J.C. Nouls, P. Mollast, J.C. Brackman, G. Van Binst, J. Pecher, and R.H. Martin, Tetrahedron Letters, 2731 (1968).